

Division of the History of Chemistry American Chemical Society

## Citation for Chemical Breakthrough



Discovery of the First Noble Gas Compound

Proceedings of the Chemical Society 1962, 6, 218.

## Xenon Hexafluoroplatinate(v) Xe<sup>+</sup>[PtF<sub>6</sub>]<sup>-</sup> By Neil Bartlett

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Tensimetric titration of xenon (AIRCO "Reagent Grade") with platinum hexafluoride has proved the existence of a 1:1 compound, XePtF<sub>6</sub>. This is an orange-yellow solid, which is insoluble in carbon tetrachloride, and has a negligible vapour pressure at room temperature. It sublimes in a vacuum when heated and the sublimate, when treated with water vapour, rapidly hydrolyses, xenon and oxygen being evolved and hydrated platinum dioxide deposited:

 $2XePtF_6 + 6H_2O \rightarrow 2Xe + O_2 + 2PtO_2 + 12HF$ 

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